ISSN 1070-4280, Russian Journal of Organic Chemistry, 2008, Vol. 44, No. 10, pp. 1486–1489. © Pleiades Publishing, Ltd., 2008. Original Russian Text © G.N. Rozentsveig, A.V. Popov, I.B. Rozentsveig, G.G. Levkovskaya, 2008, published in Zhurnal Organicheskoi Khimii, 2008, Vol. 44, No. 10, pp. 1509–1513.

> Dedicated to Full Member of the Russian Academy of Sciences B.A. Trofimov on his 70th anniversary

## **Reactions of** *N***-(2,2,2-Trichloroethylidene)- and** *N***-(2,2-Dichloro-2-phenylethylidene)arenesulfonamides with Biuret**

G. N. Rozentsveig, A. V. Popov, I. B. Rozentsveig, and G. G. Levkovskaya

Favorskii Irkutsk Institute of Chemistry, Siberian Division, Russian Academy of Sciences, ul. Favorskogo 1, Irkutsk, 664033 Russia e-mail: i roz@irioch.irk.ru

Received April 30, 2008

**Abstract**—N-(2,2,2-Trichloroethylidene)- and N-(2,2-dichloro-2-phenylethylidene)arenesulfonamides react with an equimolar amount of biuret to give 1-(1-arylsulfonylamino-2,2,2-trichloroethyl)- or 1-(1-arylsulfonyl-amino-2,2-dichloro-2-phenylethyl)biurets. The reactions with 2 equiv of N-(polychloroethylidene)arenesulfon-amides involve both amino groups in the biuret molecule, yielding the corresponding 1,5-bis(1-arylsulfonyl-amino-2,2,2-trichloroethyl)- and 1,5-bis(1-arylsulfonylamino-2,2-dichloro-2-phenylethyl)biurets.

**DOI:** 10.1134/S1070428008100138

Polyamide systems having sulfonamide or urea fragments attract interest as potential ligands, strong NH acids [1], catalysts, reagents, and substrates for asymmetric synthesis and supramolecular chemistry [2]; they exhibit biological activity [3, 4] and are used in the synthesis of macrocyclic nitrogen-containing compounds [5]. Therefore, development of efficient procedures for the preparation of functionalized polyamide systems is an important problem.

Convenient synthetic approaches to various substituted amide derivatives can be based on transformations of polyhalogenated aldehyde imines. The presence of electron-withdrawing groups in molecules of such Schiff bases considerably activates the CH=N bond toward nucleophilic reagents, which is widely used for various synthetic purposes [6].

We previously studied reactions of *N*-(polychloroethylidene)arenesulfonamides with amides [7, 8], thioamides, urea, and phenylurea [9–11] with a view to develop procedures for selective synthesis of new functionalized haloalkylamide systems. While continuing studies in this line, in the present work we examined reactions of *N*-(2,2,2-trichloroethylidene)- and *N*-(2,2-dichloro-2-phenylethylidene)arensulfonamides **Ia–Ic** and **IIa–IIc** with biuret. These reactions were expected to produce polyamide systems having sulfonamide, urea, and polyhaloalkyl fragments, which may be interesting as potential biologically active sub-



I, III, V, X = Cl; II, IV, VI, X = Ph; Ar = Ph (a), 4-ClC<sub>6</sub>H<sub>4</sub> (b), 4-MeC<sub>6</sub>H<sub>4</sub> (c).

stances, ligands, and intermediate products for subsequent transformations.

Initial Schiff bases Ia-Ic and IIa-IIc were synthesized according to the procedures developed previously, i.e., by radical reaction of the corresponding N,N-dichloroarenesulfonamides with trichloroethylene or phenylacetylene [12, 13]. Biuret reacted with Schiff bases Ia-Ic and IIa-IIc on heating the reactants to 90-100°C. The use of dimethylformamide as solvent considerably improves the yield of the addition products (the reaction mixture is homogeneous). The reactions of biuret with equimolar amounts of Schiff bases I and II involved only one amino group in the former, and the products were the corresponding 1-(1-arylsulfonylamino-2,2,2-trichloroethyl)- and 1-(1-arylsulfonylamino-2,2-dichloro-2-phenylethyl)biurets IIIa-IIIc and IVa-IVc (Scheme 1). By reactions of biuret with 2 equiv of compounds I and II we obtained 1,5-bis(1arylsulfonylamino-2,2,2-trichloroethyl)- and 1,5-bis(1arylsulfonylamino-2,2-dichloro-2-phenylethyl)biurets V and VI, respectively, as a result of condensation involving both NH<sub>2</sub> groups in the nucleophilic reagent.

Trichloroacetaldehyde derivatives **Ia–Ic** turned out to be more reactive in the examined process than the corresponding dichloro(phenyl)acetaldehyde derivatives **IIa–IIc** due to stronger electron-withdrawing effect of the trichloromethyl group as compared to dichloro(phenyl)methyl. The reaction of biuret with compounds **I** required approximately twice as short time (12 h) as analogous transformation of N-[2,2-dichloro-2-phenylethylidene)arenesulfonamides. The yield of addition products **III–VI** were 79–97%.

The structure of compounds III–VI was proved by the spectral data and elemental analyses. The IR spectra of new biuret derivatives III–VI contained absorption bands typical of NH, carbonyl, and sulfonamide groups. Reduced stretching vibration frequency of the NH and NH<sub>2</sub> groups indicates that they are involved in hydrogen bonding. Taking into account that molecules III–VI possess several proton-donor and proton-acceptor groups and that internal rotation about the formally single (O)C–N amide bonds is restricted, both interand intramolecular hydrogen bonds may be formed.

In the <sup>1</sup>H NMR spectra of addition products **III–VI** in DMSO- $d_6$ , protons in the NHCHNH fragment resonated as a doublet of doublets at  $\delta$  5.7–5.9 ppm (CH) and a doublet and a singlet from the NH protons in a weaker field ( $\delta$  8.6–9.3 ppm). Presumably, the NH group in the urea fragment is involved in tautomeric transformations; therefore, the NH signal appears as a broadened singlet despite spin–spin coupling with the methine proton. In addition, the spectra contained broadened singlets due to protons in the (O)CNHC(O) and NH<sub>2</sub> groups and signals from aromatic protons. The chemical shifts of the NCHN proton in bis-sulfonamides **Va–Vc** and **VIa–VIc** range from  $\delta$  5.0 to 5.5 ppm [12,13]. The downfield shift of the CHCCl<sub>2</sub>X proton signal in the <sup>1</sup>H NMR spectra of biuret derivatives **III–VI** is likely to result from deshielding effect of the carbonyl group. Compounds **III–VI** displayed in the <sup>13</sup>C NMR spectra signals corresponding to carbon atoms in the carbonyl and polyhalomethyl groups, NCN carbon atom, and carbon atoms in the aromatic rings.

Substituted biurets **III–VI** are colorless crystalline substances which are readily soluble in DMSO, acetone, and aqueous alkali and insoluble in water. The presence of NHC=O and C–Cl fragments in their molecules makes them promising as reagents for the synthesis of nitrogen-containing heterocyclic compounds and complexes with metal salts.

## EXPERIMENTAL

The IR spectra were recorded in KBr on a Bruker IFS-25 spectrometer. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were measured on a Bruker DPX-400 instrument at 400.61 and 100.13 MHz, respectively, using DMSO- $d_6$  as solvent and tetramethylsilane as internal reference. *N*-(2,2,2-Trichloroethylidene)arenesulfonamides Ia–Ic [12] and *N*-(2,2-dichloro-2-phenylethylidene)arenesulfonamides IIa–IIc [13] were prepared according to the procedures reported previously.

1-(2,2,2-Trichloro-1-phenylsulfonylaminoethyl)biuret (IIIa). A mixture of 2.87 g (0.01 mol) of Schiff base Ia, 1.03 g (0.01 mol) of biuret, and 5 ml of anhydrous dimethylformamide was stirred for 12 h at 90-100°C. The mixture was poured into 30-40 ml of water, and the precipitate was filtered off, dried, and washed with diethyl ether (30-40 ml) until colorless washings. Yield 3.08 g (79%), mp 220-222°C (decomp.). IR spectrum, v,  $cm^{-1}$ : 1165, 1330 (SO<sub>2</sub>); 1690, 1714 (C=O); 3025-3235 br, 3380, 3480 (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 5.85 d.d (1H, CH, <sup>3</sup>J = 8.5 Hz); 6.80 br.s (2H, NH<sub>2</sub>); 7.51 m, 7.63 m, and 7.82 m (5H, C<sub>6</sub>H<sub>5</sub>); 8.68 br.s (1H, CONHCO); 8.79 s (1H, CONHCH); 9.24 d (1H, SO<sub>2</sub>NH,  ${}^{3}J = 8.5$  Hz). <sup>13</sup>C NMR spectrum,  $\delta_{C}$ , ppm: 70.44 (CH); 100.99 (CCl<sub>3</sub>); 126.67, 128.86, 132.75, 140.73 (C<sub>arom</sub>); 153.19 (CONH<sub>2</sub>); 155.05 (NHCONH). Found, %: C 30.95; H 2.67; Cl 27.45; N 14.25; S 8.15. C<sub>10</sub>H<sub>11</sub>Cl<sub>3</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 30.83; H 2.85; Cl 27.30; N 14.38; S 8.23.

**1-[2,2,2-Trichloro-1-(4-chlorophenylsulfonylamino)ethyl]biuret (IIIb)** was synthesized in a similar way from 3.21 g (0.01 mol) of sulfonamide **Ib**. Yield 3.73 g (88%), mp 245–247°C (decomp.). IR spectrum, v, cm<sup>-1</sup>: 1150, 1340 (SO<sub>2</sub>); 1700, 1720 (C=O); 3140, 3210, 3270, 3340, 3445 (NH). <sup>1</sup>H NMR spectrum, δ, ppm: 5.82 d.d (1H, CH, <sup>3</sup>*J* = 9.5 Hz), 6.64 br.s (2H, NH<sub>2</sub>), 7.59 and 7.82 (4H, C<sub>6</sub>H<sub>4</sub>, *AA'BB'* system), 8.69 br.s (1H, CONHCO), 8.81 s (1H, CONHCH), 9.29 d (1H, SO<sub>2</sub>NH, <sup>3</sup>*J* = 9.5 Hz). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 70.35 (CH); 100.69 (CCl<sub>3</sub>); 128.61, 128.86, 137.65, 139.44 (C<sub>arom</sub>); 153.06 (CONH<sub>2</sub>); 154.96 (NHCONH). Found, %: C 28.73; H 2.41; Cl 33.86; N 13.42; S 7.44. C<sub>10</sub>H<sub>10</sub>Cl<sub>4</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 28.32; H 2.38; Cl 33.44; N 13.21; S 7.56.

1-[2,2,2-Trichloro-1-(4-methylphenylsulfonylamino)ethyl]biuret (IIIc) was synthesized in a similar way from 3.01 g (0.01 mol) of sulfonamide Ic. Yield 3.56 g (88%), mp 225-227°C (decomp.). IR spectrum, v, cm<sup>-1</sup>: 1165, 1330 (SO<sub>2</sub>); 1700, 1720 (C=O); 3100, 3135, 3200, 3340, 3455 (NH). <sup>1</sup>H NMR spectrum, δ, ppm: 2.37 s (3H, CH<sub>3</sub>), 5.81 d.d (1H, CH,  ${}^{3}J = 9.5$  Hz), 6.79 br.s (2H, NH<sub>2</sub>), 7.31 and 7.69 (4H, C<sub>6</sub>H<sub>4</sub>, AA'BB' system), 8.68 br.s (1H, CONHCO), 8.77 s (1H, CONHCH), 9.12 d (1H, SO<sub>2</sub>NH,  ${}^{3}J = 9.5$  Hz). <sup>13</sup>C NMR spectrum,  $\delta_{\rm C}$ , ppm: 21.40 (CH<sub>3</sub>); 70.79 (CH); 101.34 (CCl<sub>3</sub>); 127.09, 129.61, 138.17, 143.45 (Carom); 153.51 (CONH<sub>2</sub>); 155.39 (NHCONH). Found, %: C 32.69; H 3.28; Cl 26.49; N 13.76; S 7.88. C<sub>11</sub>H<sub>13</sub>Cl<sub>3</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 32.73; H 3.25; Cl 26.35; N 13.88; S 7.94.

1-(2,2-Dichloro-2-phenyl-1-phenylsulfonylaminoethyl)biuret (IVa). A mixture of 3.28 g (0.01 mol) of sulfonamide IIa, 1.03 g (0.01 mol) of biuret, and 5 ml of DMF was stirred for 20 h at 100°C. The mixture was then treated as described above. Yield 3.79 g (88%), mp 175–177°C. IR spectrum, v,  $cm^{-1}$ : 1150, 1330 (SO<sub>2</sub>); 1670, 1700 (C=O); 3275-3455 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 5.95 d.d (1H, CH,  ${}^{3}J = 10.0$  Hz), 6.80 br.s (2H, NH<sub>2</sub>), 7.52 m and 7.82 m (10H, C<sub>6</sub>H<sub>5</sub>), 8.53 br.s (1H, CONHCO), 8.59 s (1H, CONHCH), 8.84 d (1H, SO<sub>2</sub>NH,  ${}^{3}J = 10.0$  Hz). <sup>13</sup>C NMR spectrum,  $\delta_{\rm C}$ , ppm: 67.33 (CH); 94.98 (CCl<sub>2</sub>); 126.67, 128.15, 128.56, 128.86, 129.49, 136.98, 138.42, 139.73 (Carom); 152.68 (CONH<sub>2</sub>); 154.28 (NHCONH). Found, %: C 44.73; H 3.71; Cl 16.52; N 12.85; S 7.35. C<sub>16</sub>H<sub>16</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 44.56; H 3.74; Cl 16.44; N 12.99; S 7.43.

**1-[2,2-Dichloro-1-(4-chlorophenylsulfonylamino)-2-phenylethyl]biuret (IVb)** was synthesized in a similar way from 3.64 g (0.01 mol) of sulfonamide **Ib.** Yield 4.25 g (91%), mp 167–169°C. IR spectrum, v, cm<sup>-1</sup>: 1150, 1340 (SO<sub>2</sub>); 1675, 1700 (C=O); 3250–3400 br. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 5.96 d.d (1H, CH, <sup>3</sup>*J* = 10.3 Hz), 6.83 br.s (2H, NH<sub>2</sub>), 7.42 m and 7.95 m (9H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 8.54 br.s (1H, CONHCO), 8.57 s (1H, CONHCH), 8.86 d (1H, SO<sub>2</sub>NH, <sup>3</sup>*J* = 10.3 Hz). <sup>13</sup>C NMR spectrum,  $\delta_{C}$ , ppm: 68.23 (CH); 94.60 (CCl<sub>2</sub>); 126.97, 128.21, 128.61, 128.68, 129.58, 137.32, 138.54, 139.80 (C<sub>arom</sub>); 152.87 (CONH<sub>2</sub>), 155.01 (NHCONH). Found, %: C 41.38; H 3.35; Cl 22.74; N 12.12; S 6.75. Cl<sub>6</sub>H<sub>15</sub>Cl<sub>3</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 41.26; H 3.25; Cl 22.84; N 12.03; S 6.88.

1-[2,2-Dichloro-1-(4-methylphenylsulfonylamino)-2-phenylethyl|biuret (IVc) was synthesized in a similar way from 3.42 g (0.01 mol) of sulfonamide **IIc**. Yield 3.65 g (82%), mp 196–198°C. IR spectrum, v, cm<sup>-1</sup>: 1160, 1325 (SO<sub>2</sub>); 1660, 1695 (C=O); 3275-3440 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.33 s (3H, CH<sub>3</sub>), 5.96 d.d (1H, CH,  ${}^{3}J = 9.8$  Hz), 6.67 br.s (2H, NH<sub>2</sub>), 7.22 m and 7.66 m (9H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 8.37 br.s (1H, CONHCO), 8.52 s (1H, CONHCH), 8.65 d (1H, SO<sub>2</sub>NH,  ${}^{3}J = 9.8$  Hz).  ${}^{13}$ C NMR spectrum,  $\delta_{C}$ , ppm: 21.07 (CH<sub>3</sub>); 68.27 (CH); 94.88 (CCl<sub>2</sub>); 126.76, 127.07, 128.25, 129.09, 129.61, 138.20, 138.77, 142.76 (Carom); 152.99 (CONH<sub>2</sub>); 155.07 (NHCONH). Found, %: C 45.79; H 3.99; Cl 15.78; N 12.67; S 7.28. C<sub>17</sub>H<sub>18</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>4</sub>S. Calculated, %: C 45.85; H 4.07; Cl 15.92; N 12.58; S 7.20.

**1,5-Bis(2,2,2-trichloro-1-phenylsulfonylaminoethyl)biuret (Va).** A mixture of 2.87 g (0.01 mol) of sulfonamide **Ia**, 0.52 g (0.005 mol) of biuret, and 5 ml of DMF was stirred for 12 h at 90–100°C. The mixture was then treated as described above. Yield 2.85 g (84%), mp 247–249°C. IR spectrum, v, cm<sup>-1</sup>: 1170, 1345 (SO<sub>2</sub>); 1655, 1690–1715 (C=O); 3250 br (NH). <sup>1</sup>H NMR spectrum, δ, ppm: 5.71 d (2H, CH, <sup>3</sup>*J* = 9.6 Hz), 7.40 m and 7.94 m (10H, C<sub>6</sub>H<sub>5</sub>), 8.89 s (1H, CONHCO), 9.23 d (2H, SO<sub>2</sub>NH, <sup>3</sup>*J* = 9.6 Hz). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 70.40 (CH); 100.69 (CCl<sub>3</sub>); 126.68, 128.88, 132.91, 140.39 (C<sub>arom</sub>); 162.50 (C=O). Found, %: C 32.48; H 2.56; Cl 31.68; N 10.78; S 9.68. C<sub>18</sub>H<sub>17</sub>Cl<sub>6</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 31.97; H 2.53; Cl 31.46; N 10.36; S 9.48.

**1,5-Bis[2,2,2-trichloro-1-(4-chlorophenylsulfonylamino)ethyl]biuret (Vb)** was synthesized in a similar way from 3.21 g (0.01 mol) of sulfonamide **Ib**. Yield 3.82 g (90%), mp 252–254°C (decomp.). IR spectrum, v, cm<sup>-1</sup>: 1170, 1345 (SO<sub>2</sub>); 1670 br (C=O); 3275–3380 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 5.72 d (2H, CH, <sup>3</sup>J = 8.0 Hz), 7.51 and 7.75 (8H, C<sub>6</sub>H<sub>4</sub>, *AA'BB'* system), 8.99 s (1H, CONHCO), 9.34 d (2H, SO<sub>2</sub>NH,  ${}^{3}J$  = 8.0 Hz).  ${}^{13}$ C NMR spectrum,  $\delta_{C}$ , ppm: 71.57 (CH); 101.68 (CCl<sub>3</sub>); 129.85, 130.13, 139.00, 140.50 (C<sub>arom</sub>); 153.71 (C=O). Found, %: C 29.22; H 2.09; Cl 38.45; N 9.25; S 8.45. C<sub>18</sub>H<sub>15</sub>Cl<sub>8</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 29.02; H 2.03; Cl 38.07; N 9.40; S 8.61.

**1,5-Bis[2,2,2-trichloro-1-(4-methylphenylsulfonylamino)ethyl]biuret (Vc)** was synthesized in a similar way from 3.01 g (0.01 mol) of sulfonamide **Ic**. Yield 3.65 g (86%), mp 236–238°C. IR spectrum, v, cm<sup>-1</sup>: 1170, 1340 (SO<sub>2</sub>); 1680 br (C=O); 3250– 3320 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.28 s (6H, CH<sub>3</sub>), 5.74 d (2H, CH, <sup>3</sup>*J* = 9.9 Hz), 7.23 and 7.62 (8H, C<sub>6</sub>H<sub>4</sub>, *AA'BB'* system), 8.88 s (1H, CONHCO), 9.05 d (2H, SO<sub>2</sub>NH, <sup>3</sup>*J* = 9.9 Hz). <sup>13</sup>C NMR spectrum,  $\delta_{C}$ , ppm: 20.49 (CH<sub>3</sub>); 71.04 (CH); 100.98 (CCl<sub>3</sub>); 127.55, 128.92, 137.68, 140.25 (C<sub>arom</sub>); 153.53 (C=O). Found, %: C 34.38; H 3.09; Cl 30.28; N 9.86; S 9.25. C<sub>20</sub>H<sub>21</sub>Cl<sub>6</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 34.11; H 3.01; Cl 30.20; N 9.94; S 9.10.

1,5-Bis(2,2-dichloro-2-phenyl-1-phenylsulfonylaminoethyl)biuret (VIa). A mixture of 3.28 g (0.01 mol) of sulfonamide IIa, 0.52 g (0.005 mol) of biuret, and 5 ml of DMF was stirred for 20 h at 100°C. The mixture was then treated as described above. Yield 3.50 g (92%), mp 85–87°C. IR spectrum, v,  $cm^{-1}$ : 1160, 1340 (SO<sub>2</sub>); 1655, 1700 (C=O); 3275-3360 br (NH). <sup>1</sup>H NMR spectrum, δ, ppm: 5.94 d.d (2H, CH,  $^{3}J = 9.6$ , 10.2 Hz), 7.35 m and 7.95 m (20H, C<sub>6</sub>H<sub>5</sub>), 8.58 s (1H, CONHCO), 8.83 d (2H, CONHCH,  ${}^{3}J =$ 9.6 Hz), 8.85 d (2H, SO<sub>2</sub>NH,  ${}^{3}J = 10.2$  Hz).  ${}^{13}C$  NMR spectrum, δ<sub>C</sub>, ppm: 68.21 (CH); 94.43 (CCl<sub>2</sub>); 126.83, 127.22, 127.89, 128.13, 128.20, 128.86, 128.99, 129.85 (C<sub>arom</sub>); 152.79 (C=O). Found, %: C 47.56; H 3.48; Cl 18.55; N 9.25; S 8.25. C<sub>30</sub>H<sub>27</sub>Cl<sub>4</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 47.44; H 3.58; Cl 18.67; N 9.22; S 8.44.

**1,5-Bis[2,2-dichloro-1-(4-chlorophenylsulfonylamino)-2-phenylethyl]biuret (VIb)** was synthesized in a similar way from 3.64 g (0.01mol) of sulfonamide **IIb.** Yield 4.04 g (97%), mp 78–80°C. IR spectrum, v, cm<sup>-1</sup>: 1170, 1340 (SO<sub>2</sub>); 1660, 1700 (C=O); 3280– 3370 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 5.95 d.d (2H, CH, <sup>3</sup>*J* = 9.5, 10.0 Hz), 7.39 m and 7.95 m (18H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 8.58 s (1H, CONHCO), 8.83 d (2H, CONHCH, <sup>3</sup>*J* = 9.5 Hz), 8.86 d (2H, SO<sub>2</sub>NH, <sup>3</sup>*J* = 10.0 Hz). <sup>13</sup>C NMR spectrum,  $\delta_{\rm C}$ , ppm: 69.26 (CH); 94.60 (CCl<sub>2</sub>); 126.94, 127.33, 127.55, 127.97, 128.00, 128.76, 129.00, 129.58 (C<sub>arom</sub>); 152.84 (C=O). Found, %: C 43.65; H 3.09; Cl 25.46; N 8.39; S 7.78. C<sub>30</sub>H<sub>25</sub>Cl<sub>6</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 43.50; H 3.04; Cl 25.68; N 8.45; S 7.74. **1,5-Bis[2,2-dichloro-1-(4-methylphenylsulfonylamino)-2-phenylethyl]biuret (VIc)** was synthesized in a similar way from 3.42 g (0.01 mol) of sulfonamide **IIc.** Yield 3.51 g (89%), mp 82–84°C. IR spectrum, v, cm<sup>-1</sup>: 1165, 1340 (SO<sub>2</sub>); 1660 (C=O); 3275–3370 br (NH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.34 s (6H, CH<sub>3</sub>); 5.93 d.d (2H, CH, <sup>3</sup>J = 9.4, 10.1 Hz), 7.32 m and 7.98 m (18H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 8.49 s (1H, CONHCO), 8.82 d (2H, CONHCH, <sup>3</sup>J = 9.4 Hz), 8.87 d (2H, SO<sub>2</sub>NH, <sup>3</sup>J = 10.1 Hz). <sup>13</sup>C NMR spectrum,  $\delta_{\rm C}$ , ppm: 24.73 (CH<sub>3</sub>); 68.85 (CH); 94.58 (CCl<sub>2</sub>); 124.99, 127.03, 127.45, 127.99, 128.08, 128.67, 129.12, 129.62 (C<sub>arom</sub>); 152.96 (C=O). Found, %: C 48.85; H 3.95; Cl 18.11; N 8.95; S 8.05. C<sub>32</sub>H<sub>31</sub>Cl<sub>4</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>. Calculated, %: C 48.80; H 3.97; Cl 18.01; N 8.89; S 8.14.

## REFERENCES

- 1. Ackermann, L., Bergman, R.G., and Loy, R.N., J. Am. Chem. Soc., 2003, vol. 125, p. 11956.
- 2. Evans, D.A. and Nelson, S.G., J. Am. Chem. Soc., 1997, vol. 119, p. 6452.
- Pena, C., Alfonso, I., Nicolas, H., and Gotor, V., *Tetra*hedron Lett., 2005, vol. 46, p. 2783.
- Boss, C., Bolli, M.H., Weller, T., Fischli, W., and Clozel, M., *Bioorg. Med. Chem. Lett.*, 2003, vol. 13, p. 951.
- Alfonso, I., Rebolledo, F., and Gotor, V., *Chem. Eur. J.*, 2000, vol. 6, p. 3331.
- 6. Levkovskaya, G.G., Drozdova, T.I., Rozentsveig, I.B., and Mirskova, A.N., *Usp. Khim.*, 1999, vol. 68, p. 638.
- Rozentsveig, I.B., Ushakova, I.V., Kondrashov, E.V., Rozentsveig, G.N., Levkovskaya, G.G., and Mirskova, A.N., *Russ. J. Org. Chem.*, 2005, vol. 41, p. 1558.
- 8. Evstaf'eva, I.T., Sarapulova, G.I., Levkovskaya, G.G., and Aizina, J.A., *Arkivoc*, 2003, part (xiii), p. 45.
- 9. Rozentsveig, G.N., Aizina, Yu.A., Rozentsveig, I.B., Levkovskaya, G.G., Sarapulova, G.I., Mirskova, A.N., and Drozdova, T.I., *Russ. J. Org. Chem.*, 2003, vol. 39, p. 554.
- Mirskova, A.N., Drozdova, T.I., Levkovskaya, G.G., Bannikova, O.B., Kalikhman, I.D., and Voronkov, M.G., *Zh. Org. Khim.*, 1982, vol. 18, p. 1407.
- Mirskova, A.N., Drozdova, T.I., Levkovskaya, G.G., Gogoberidze, I.T., Ochirov, Yu.D., Zarubina, V.N., Zhovtyi, I.F., and Voronkov, M.G., *Fiziologicheski aktivnye veshchestva* (Physiologically Active Substances), Kiev: Naukova Dumka, 1989, p. 84.
- Mirskova, A.N., Levkovskaya, G.G., Drozdova, T.I., Kalikhman, I.D., and Voronkov, M.G., *Zh. Org. Khim.*, 1982, vol. 18, p. 452.
- 13. Drozdova, T.I., Levkovskaya, G.G., and Mirskova, A.N., *Zh. Org. Khim.*, 1992, vol. 28, p. 1236.

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 44 No. 10 2008